Nakijinamines C–E, New Heteroaromatic Alkaloids from the Sponge *Suberites* Species

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Three new heteroaromatic alkaloids, nakijinamines C-E(1-3), which are a hybrid of the aaptamine-type and bromoindole alkaloids possessing a taurine- or histidine-derived residue, have been isolated from an Okinawan marine sponge *Suberites* species. The structures of 1–3 were elucidated on the basis of spectroscopic data and chemical conversions. Nakijinamines C(1) and E(3) are the first natural products possessing a 1*H*-oxazolo[4',5':4,5]benzo[1,2,3-*de*][1,6]naphthyridine ring system.

Marine sponges have been recognized as a rich source of interesting bioactive metabolites with a unique structure.¹ During our continuing search for secondary metabolites possessing a unique structure from marine sponges,² we have investigated the extract of an Okinawan marine sponge *Suberites* sp. (SS-1084) and isolated three new heteroaromatic alkaloids, nakijinamines C-E (1–3), which are a hybrid of the aaptamine-type and bromoindole alkaloids possessing a taurine- or histidine-derived residue. In particular, nakijinamine C (1) and E (3) had a unique tetracyclic ring system containing a 2,4,5,6,7-pentasubstituted benzoxazole ring. Here we describe the isolation and structure elucidation of 1–3.

The sponge *Suberites* sp. (SS-1084, 0.4 kg wet weight) was extracted with MeOH. The extract was partitioned

between EtOAc and H₂O, and then the aqueous layer was extracted with *n*-BuOH. The *n*-BuOH-soluble materials were subjected to C₁₈ column chromatography followed by repeated reversed-phase HPLC to afford nakijinamines C (1, 37.0 mg, 9.3×10^{-30} %, wet weight),³ D (2, 11.2 mg,

(4) Because of limited solubility, physicochemical properties of nakijinamine D (2) were obtained from a methylated derivative 6 and an elimination derivative 7 (see the Supporting Information).

(5) Nakijinamine E (3): orange amorphous solid; $[\alpha]^{21}_{D} - 5$ to +2 (c 0.15, MeOH); UV (0.1 M HCl aq, pH 1) λ_{max} 224 (log ε 4.4), 247 (4.2), 286 (4.0 sh), 294 (4.1), 306 (4.0 sh), 398 (3.7), and 456 nm (3.7); IR (film) ν_{max} 3400, 3222, 3086, 2884, 1656, 1525, 1449, 1299, 892, and 822 cm⁻¹; ¹H and ¹³C NMR (DMSO-d₆) see Table 1; ESIMS (pos) m/z 596/598 (1:1, $[M-H]^+$); HRESIMS (pos) m/z 596.14061 ($[M-H]^+$, calcd for C₃₀H₂₇⁻⁷⁹BrN₇O₂, 596.14041).

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⁽³⁾ Nakijinamine C (1): yellow amorphous solid; $[\alpha]^{21}{}_{\rm D}$ -3 to +5 (*c* 0.15, MeOH); UV (0.1 M HCl aq, pH 1) $\lambda_{\rm max}$ 224 (log ε 4.6), 240 (4.5 sh), 256 (4.3), 284 (3.9 sh), 294 (3.9), 338 (4.1), 375 (3.9 sh), 406 (3.6 sh), and 432 nm (3.2 sh); IR (film) $\nu_{\rm max}$ 3384, 2997, 2844, 1661, 1616, 1550, 1472, 1450, 1238, 1207, 1175, 1034, and 804 cm⁻¹; ¹H and ¹³C NMR (DMSO-*d*₆) see Table 1; ESIMS (pos) *m*/*z* 582/584 (1:1, [M]⁺); HRE-SIMS (pos) *m*/*z* 582.08073 ([M]⁺, calcd for C₂₆H₂₅⁻⁷BrN₅O₄S, 582.08051).

			1					3	
position	$\delta_{\rm C}{}^c$ (mult.)	$\delta_{\mathrm{N}}{}^{b,c}$	$\delta_{ m H}{}^c$ (mult., J in Hz)	${}^{1}J_{\mathrm{CH}}{}^{c}$ (Hz)	HMBC ^c (H to C or N)	position	$\delta_{\rm C}{}^c$ (mult.)	$\delta_{ m H}{}^c$ (mult., J in Hz)	$\begin{array}{l} \mathrm{HMBC}^{d} \\ (\mathrm{H \ to \ C}) \end{array}$
1		145.5	13.73(brs)			1			
2	$141.1\mathrm{CH}$		7.88(d, 6.9)	179.4	1, 3a, 9a	2	$141.5\mathrm{CH}$	8.29(d, 6.9)	3, 3a, 9a
3	$101.2\mathrm{CH}$		6.63(d, 6.9)	169.2	1, 2, 9b	3	$102.7\mathrm{CH}$	6.91 (d, 6.9)	2, 9b
3a	$151.1\mathrm{C}$					3a	$151.0\mathrm{C}$		
4		143.5	13.00(brs)			4		13.36(brs)	
5	$130.3\mathrm{CH}$		7.54(d, 7.3)	179.4	3a, 6, 6a	5	$131.7\mathrm{CH}$	7.72(d, 7.6)	3a, 6, 6a
6	$109.6\mathrm{CH}$		7.59^g	159.6	4	6	$109.6\mathrm{CH}$	$7.82(\mathrm{brs})^h$	5, 7, 9b
6a	$130.1\mathrm{C}$					6a	$132.4~\mathrm{C}$		
7	$110.9\mathrm{C}$					7	$111.8\mathrm{C}$		
8	$152.4\mathrm{C}$					8	$152.2\mathrm{C}$		
9	$125.8\mathrm{C}$					9	$125.4\mathrm{C}$		
9a	$131.8\mathrm{C}$					9a	$133.9\mathrm{C}$		
9b	119.0 C					9b	$119.2~\mathrm{C}$		
1'		135.8	11.38(d, 2.0)		2′, 3′, 3′a, 7′a	1'		11.45(brd, 1.8)	3′, 3′a, 7′a
2'	$125.4\mathrm{CH}$		7.57 ^g	179.4	1′, 3′, 3′a, 7′a	2'	$125.2~\mathrm{CH}$	7.69(d, 1.8)	3′, 3′a, 7′a
3'	$112.8~\mathrm{C}$					3'	$112.9\mathrm{C}$		
3′a	$124.5\mathrm{C}$					3′a	$124.6\mathrm{C}$		
4'	$120.7\mathrm{CH}$		7.83(brd, 8.6)	159.0	6′, 7′a	4'	$120.6\mathrm{CH}$	$7.60(\mathrm{brs})^h$	3′, 6′, 7′a
5'	$122.0~\mathrm{CH}$		7.14(brd, 8.6)	159.0	3'a, 7'	5'	$121.8\mathrm{CH}$	7.07(d, 8.8)	3'a, 6', 7'
6′	$114.3\mathrm{C}$					6′	$114.3\mathrm{C}$		
7'	$114.2\mathrm{CH}$		7.51(d, 1.3)	159.6	3'a, 5', 6'	7'	$114.2\mathrm{CH}$	7.52(d, 1.4)	3′a, 5′, 6′, 7′a
7′a	136.9 C					7′a	$137.1\mathrm{C}$		
$1^{\prime\prime}$ -NMe	$52.9 \ \mathrm{CH}_3$	51.0	$3.08^{e}(s)$		NMe	1''-NMe	$52.9\mathrm{CH}_3$	$3.15^{e}(s)$	1''
$1^{\prime\prime}$	$67.1\mathrm{CH}_2$		$4.75({\rm dd},13.8,9.1)$	149.4		$1^{\prime\prime}$	$67.8\mathrm{CH}_2$	$4.55({\rm dd},13.7,4.5)$	7, 2′, 2′′, 1′′-NMe
			4.26(dd, 13.8, 4.0)	139.2	7			4.46 (dd, 13.7, 6.6)	7, 2′, 2′′, 1′′-NMe
$2^{\prime\prime}$	$29.8\mathrm{CH}$		5.43(dd, 9.1, 4.0)	119.4	6a, 7, 8, 2', 3'	$2^{\prime\prime}$	$29.4\mathrm{CH}$	5.48(dd, 5.6, 5.6)	6a, 7, 8, 2', 3', 1''
1′′′′-N		238.0							
1'''	$161.6\mathrm{C}$					$1^{\prime\prime\prime}$	$156.9\mathrm{C}$		
2'''	$51.2\mathrm{CH}_2$		4.45 (d, 13.6)	140^g	2''', 3'''	$2^{\prime\prime\prime}$	$170.2\mathrm{C}$		
			4.39(d, 13.6)	140^g	2''', 3'''				
						3'''	$137.2\mathrm{C}$		
						$5^{\prime\prime\prime}$	$141.0~\mathrm{CH}$	8.07(s)	
						7'''	$132.9~\mathrm{CH}$	8.67(s)	3''', 6'''-Me
						6′′′-Me	$33.8\mathrm{CH}_3$	$3.85^{f}(s)$	5′′′, 7′′′

Table 1. ¹H, ¹³C, and ¹⁵N NMR Data of Nakijinamines C (1) and E (3) in DMSO- d_6^a

 a ¹H and 13 C NMR spectra were recorded at 600 and 150 MHz, respectively. b Assigned from 1 H $^{-15}$ N HMBC spectrum (500 MHz). c Recorded at 300 K. d Recorded at 350 K. e 9H. f 3H. g J values were not determined correctly because of overlap with other signal(s). h Observed as a broad signal because of restricted rotation.

 2.8×10^{-3} %),⁴ and E (3, 5.0 mg, 1.3×10^{-3} %)⁵ together with known alkaloids, bisdemethylaaptamine,⁶ and aaptamine.⁷

Nakijinamine C (1) was obtained as an optically inactive yellow amorphous solid with limited solubility, which was soluble in DMSO and slightly soluble in MeOH. The molecular formula, $C_{26}H_{24}BrN_5O_4S$, was established by HRESIMS (m/z 582.08073 [M]⁺, Δ -0.22 mmu). IR absorptions (ν_{max} 3384, 1175, and 1034 cm⁻¹) implied the presence of OH and/or NH and sulfonate

functionalities, and a conjugated aromatic chromophore was suggested by UV absorptions (λ_{max} 256 and 338 nm) under acidic conditions. ¹H NMR (Table 1) spectrum of **1** in DMSO-*d*₆ at 300 K included 3 D₂O-exchangeable, 8 sp², and 14 sp³ protons, while 12 sp² quaternary carbons, 8 sp² methines, 1 sp³ methine, 2 sp³ methylenes, and 3 sp³ methyls were observed in the ¹³C NMR (Table 1) spectrum of **1**.

 ${}^{1}\text{H}-{}^{1}\text{H}$ COSY, ${}^{1}\text{H}-{}^{13}\text{C}$ and ${}^{1}\text{H}-{}^{15}\text{N}$ HMBC, and ROESY spectra of nakijinamine C (1) and the comparison of ${}^{1}\text{H}$ and ${}^{13}\text{C}$ NMR spectra and ${}^{3}J_{\text{HH}}$ and ${}^{1}J_{\text{CH}}$ values of 1 with those of bisdemethylaaptamine⁶ and aaptamine⁷ revealed the presence of a 1*H*-benzo[*de*][1,6]naphthyridine (N-1–C-9b), a 3-substituted 6-bromoindole (N-1'–C-7'a),

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and a 2,2-disubstituted N,N,N-trimethylethanaminium moiety (Figure 1). H-2" showed HMBC cross-peaks for C-6a, C-7, C-8, C-2', and C-3' and ROESY correlations for H-6 and H-4', revealing that an N-1-C-9b portion was connected to an N-1'-C-7'a portion at C-7 and C-3' via C-2". On the other hand, the connectivity of 1""-N-C-2"" was deduced by HMBC correlations from H_2-2''' to C-1''' and 1^{'''}-N. In addition, the chemical shifts of 1^{'''}-N (δ_N 238.0), C-1^{'''} ($\delta_{\rm C}$ 161.6), and CH₂-2^{'''} ($\delta_{\rm H}$ 4.45 and 4.39; $\delta_{\rm C}$ 51.2) suggested that C-1" and 1"-N were connected to each other by a double bond and that an oxygen atom and a sulfonate group were attached to C-1" and C-2", respectively. Considering the molecular formula of 1, it was implied that C-1" was connected to both C-8 and C-9 through an oxygen atom or a nitrogen atom, respectively. The ¹³C chemical shift of C-8 in 1 was close to that of aaptamine, while the ¹³C chemical shift of C-9 in 1 was observed approximately 5 ppm higher than that of aaptamine.⁷ These comparisons showed that an oxygen atom and a nitrogen atom were connected to C-8 and C-9, respectively, to form a 2,4,5,6,7-pentasubstituted benzoxazole ring (C-6a, C-7, C-8, C-9, C-9a, C-9b, 8-O, C-1", and 1^{'''}-N). Thus, the gross structure of nakijinamine C was elucidated to be 1.



Figure 1. Selected 2D NMR correlations for nakijinamine C (1) in DMSO- d_6 at 300 K.

To confirm the gross structure of nakijinamine C (1), chemical conversions of 1 were carried out (Figure 2). Compound 1 was treated with methyl iodide under basic conditions to yield 1,4-dimethylnakijinamine C (4), and successively, 4 was heated under basic conditions to give an elimination derivative (5) of 4. The ¹H NMR spectrum of 5 in DMSO- d_6 at 300 K gave sharp signals compared to those of 1 and 4, implying that a part of C–C single bond rotations in 1 and 4 were slow due to interference by three bulky substituents attached to an sp³ methine (CH-2").

Nakijinamine D (2) was obtained as a pale yellow powder with very limited solubility. Since 2 was practically insoluble in general NMR solvents, the structure of 2 was elucidated using 1,4-dimethylnakijinamine D (6) and an dihydrofuran derivative (7) of 6, which were obtained from methylation of **2** by methyl iodide and heating of **6** under basic conditions, respectively.

In the ¹H NMR spectrum of 1,4-dimethylnakijinamine D (6), very complicated signals were obtained in DMSO- d_6 at 300 K, while a part (H-5, H-6, H-4', H-5', H-1", and H-2") of the broadening signals were observed in CD₃OD at 300 K. On the other hand, the ¹H NMR spectrum of 7 in DMSO- d_6 at 300 K showed a part (H-6, H-4', H-5', and H-2") of the broadening signals; however, all signals became sharp in DMSO- d_6 at 350 K. The structures of 6 and 7 were elucidated as shown in Figure 2 on the basis of spectroscopic data of 6 and 7. Thus, the gross structure of nakijinamine D was elucidated to be 2.

Subsequently, nakijinamine C (1) and 1,4-dimethylnakijinamine D (6) were converted into 6 and 1,4-dimethylnakijinamine C (4), respectively (Figure 2), to confirm the structure relationship of nakijinamines C (1) and D (2). Compound 1 was treated under acidic conditions followed by methylation by methyl iodide to afford 6, while 6 was treated with trifluoroacetic acid to give 4. The ¹H NMR and MS data and retention times of 6 and 4 derived from 1 and 6 were identical with those derived from 1 and 2, respectively; therefore, the structure of nakijinamine D was concluded to be 2.

Nakijinamine E (**3**) was obtained as an orange amorphous solid with a molecular formula of $C_{30}H_{27}BrN_7O_2$ (*m*/*z* 596.14061 [M-H]⁺, Δ +0.20 mmu). The existence of OH and/or NH, carbonyl group(s), and a conjugated aromatic chromophore was indicated from IR and UV absorptions of **3**, while ¹H and ¹³C NMR spectra of **3** included 14 sp² quaternary carbons, 10 sp² methines, 1 sp³ methine, 1 sp³ methylene, and 4 sp³ methyls.

¹H–¹H COSY, TOCSY, HMBC, and ROESY spectra and the comparison of the chemical shifts of **3** with those of **1** indicated that **3** has the same partial structure (N-1–C-1^{'''}) as **1** (Figure 3). In addition, the presence of a 4-substituted imidazole ring was deduced by the large ¹J_{CH} values of CH-5^{'''} (209 Hz) and CH-7^{'''} (190 Hz) and HMBC correlations of H-5^{'''}/C-3^{'''}, H-5^{'''}/C-7^{'''}, and H-7^{'''}/C-3^{'''}. A methyl group (N-6^{'''}-Me, δ_H 3.85, δ_C 33.8) shifted to relatively low field showed HMBC crosspeaks to C-5^{'''} and C-7^{'''}, suggesting the attachment of the methyl group at N-6^{'''}. Considering the molecular formula of **3** and the existence of a carbonyl group (C-2^{'''}, δ_C 170.2), it was indicated that C-1^{'''} was connected to C-3^{'''} through the carbonyl group (C-2^{'''}). Thus, the gross structure of nakijinamine E was elucidated to be **3**.

Nakijinamines C-E(1-3) were obtained as an optically inactive mixture, although they have a chiral center (C-2^{''}). Optical resolution of 1 by HPLC with a chiral column

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Figure 2. Chemical conversions of nakijinamines C (1) and D (2) into 4-7 and key 2D NMR correlations for 4-7.



Figure 3. Selected 2D NMR correlations for nakijinamine E (3) in DMSO- d_6 .

resulted in the separation of two enantiomers, the ratio of which was approximately 1:1. Furthermore, each enantiomer of **1** showed opposite $[\alpha]_D$ signs and close magnitudes. Thus, **1** was suggested to be a racemate. In addition, the enantiomers of **2** and **3** could be separated by chiral HPLC, indicating both **2** and **3** were racemates as well as **1**.

To the best of our knowledge, nakijinamines C (1) and E (3) are the first natural products possessing a 1*H*-oxazole-[4',5':4,5]benzo[1,2,3-de][1,6]naphthyridine ring system. In addition, nakijinamines C–E (1–3) are the first aaptamine-type alkaloids possessing an indole moiety, although

about 23 aaptamine-type alkaloids have been isolated so far. $^{\rm 8}$

Both nakijinamines C and E (1 and 3) showed antifungal activity against *Aspergillis niger* with MIC values of $16 \,\mu g/mL$. Nakijinamines C–E (1–3) did not show cytotoxicity (IC₅₀ > 10 $\mu g/mL$) against murine leukemia P388 and L1210 cells and human epidermoid carcinoma KB cells in vitro. Further biological investigations of 1–3 are in progress.

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Supporting Information Available. Detailed experimental section and 1D and 2D NMR data for nakijinamines C and D and their derivatives. This material is available free of charge via the Internet at http://pubs.acs.org.